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PTEROPODINE AND ISOPTEROPODINE, THE ALKALOIDS FROM

UNCARIA PTEROPODA

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<u>Mitragyna</u> and <u>Uncaria</u> species of the <u>Rubiaceae</u> family contain a number of oxindole alkaloids. Recent investigations have led to the isolation and elucidation of the structure of mitraphylline as (1) from the bark of <u>Mitragyna rubrostipulacea</u>.¹ Uncarine-A and its stereoisomer, uncarine-B were isolated from the root, bark, and stem of <u>Uncaria Kawakamii</u> Hayata.² The same structure (1) has been proposed for these alkaloids.³







931

The genus of Uncaria comprises some thirty species of trees found in Malaya. The presence of alkaloids in Uncaria pteropoda collected in Ulu Langat, Selangor, Malaya, has been noted in the course of a continuing screening program for alkaloid-bearing plants.⁴ We have now examined the alkaloids of Uncaria pteropoda. Extracting the stem, bark, and root by the conventional proceedures afforded a mixture of basic material amounting to 0.4%, 2.8% and 0.94% of the respective dry weights. Paper chromatographic tests showed the presence of two alkaloids. A convenient but not complete separation of the two alkaloids can be achieved by partitioning the mixture between benzene and 0.2N acetic acid. The major alkaloid for which we propose the name pteropodine is obtained from the acid fraction, while the benzene layer yields the other base designated as isopteropodine. Both the bases are not identical with mitraphylline or uncarine-A or -B. However, from the spectral evidence, they are most probably stereoisomeric with these alkaloids.

The properties of these two alkaloids are recorded in Table I.

TABLE I

	pteropodine	isopteropodine
n. p.	217-219 [°] C	209–211 ⁰ C
A719	-102.5(c,1.0 in CHCl ₃)	-111.0(c, 1.0 in CHCl ₃)
pKa	4.8	4.05
M.W. `	-	380
picrate, m.p.	143-144°C	147-148°C
methiodide, m.p.	209-211°(softens at 203°)	217-219 ⁰ (softens at 210 ⁰)
hydrochloride,m.p.	-	181-183°C
hydroiodide, m.p.	-	216-218°C

Analyses of pteropodine and isopteropodine show that they each contain one methoxyl group, but no N-methyl group, and have molecular formula, $C_{21}H_{24}O_{4}N_{2}$. Titrations of pteropodine and isopteropodine show them to be monoacidic bases. Clearly, pteropodine and isopteropodine are stereoisomers, and this evidence is confirmed below.

Analyses	с	н	N	OMe	MW
pteropodine isopteropodine C21H24O4N2 requires	68.86 68.53 68.46	6.80 6.74 6.57	7.36 7.25 7.48	8.52 8.52 8.29(1)	380 368

The ultraviolet and infrared spectra of pteropodine and isopteropodine are similar to each other and are closely similar to those of mitraphylline¹ and uncarine-A and -B,⁵ <u>(</u>Table II and IIIa and IIIb₇, indicating that pteropodine and isopteropodine contain an oxindole chromophore and the CH₃OOC - C¹ = CHOR grouping. In support of this, the infrared peaks of pteropodine and isopteropodine at 1627 cm⁻¹ and 1627 cm⁻¹ respectively confirmed the presence of the carbomethoxyl group as discussed by Marion et al. ¹

TABLE II

<u>Ultraviolet Spectra of Pteropodine, Isopteropodine,</u> <u>Mitraphylline,¹ Uncarine-A and -B⁵ in EtOH</u>

	λ min (mμ	log E	入 max (mµ)	log C	Shoulder (mµ4)	log E
Pteropodine	225	3.93	246	4.20	280	3.25
Isopteropodine	225	4.00	246	4.22	280	3.27
Mitraphylline	224	4.02	243	4.22	280	3.09
Uncarine_A	2 25	4.06	245	4.24	278	3.09
Uncarine_B	225	4.03	244	4.24	278	3.09

934

	TABLE III	<u>1</u>	
Infrared Spec	tra of Pteropodine, Ison Chloroform (e	pteropodine, Mitraphy zm-1)	lline ^l in
	Pteropodine	Isopteropodine	Mitraphylline
NH	3415 (20)	3415 (19)	3415 (20)
C=O	1708 (93)	1708 (92)	1715 (81)
\bigcirc	1627 (76)	1627 (74)	1625 (68)

TABLE IITD

Infrared Spectra of Pteropodine, Isopteropodine, Mitraphylline, Uncarine-A and -B⁵ in Nujol mull (cm-1)

	Pteropodine	Isopteropodine	Mitraphy- lline	Uncarine-A	-B
	3446	3436	3260	3340	3200
Ester C=O	1719	1712	1725	1715	1730
Oxindole C=O	1688	1677	1704	1693	1693
\bigcirc	1627	1627	1626	1626	1626
Cyclic ether	1061	1077	1105	1107	1107
\bigcirc	750	755	755	757	757

Apart from the similarities in ultraviolet and infrared spectra, the N.M.R. spectra of pteropodine and isopteropodine show many common features, again suggesting that the two are structurally closely related - (Table IV). In the spectra of pteropodime and isopteropodime, the $-C-0CH_3$, $-C=C-0-H_3$

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TABLE	

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N.N.R. Spectra of Pteropodine, Isopteropodine, Speciofoline, Stipulatine? Careparaubine

and Isoreserpiline⁸ (in CDCl3 at 60 Mc in p.p.m. from TMS)

Function <u>al</u> Groups	Pteropodine	Isopteropodine	Speciofoline	Stipulatine	Garapangubine	Isoreserpiline
CH ₃ CHD	1.60(doublet)	1.65(doublet)			1.40(doublet)	1.37(doublet)
CH ₃ COO	3.80(singlet)	3.82(singlet)	3.78(singlet)	3.70(singlet)	3.61(singlet)	3.72(singlet)
Aromatic H	7.30(multiplet)7.28(multiplet)	7.08(triplet)	7.45(triplet)	6.55, 6.74 (einglete)	6.77, 6.90 (simelets)
Olefinic	7.70(singlet)	7.72(singlet)	7.40(singlet)	7.23(singlet)	7.44 (singlet)	7.57(singlet)
ן ב-2 ו	9.47(singlet)	9.49(singlet)	8.48(singlet)	9.16(singlet)	8.73(singlet)	7.95(singlet)

and aliphatic -NH protons give rise to singlets at 3.82 p.p.m., 7.70 p.p.m. and 9.47 p.p.m., respectively, similar to those of stipulatine and speciofoline whose structures (2) have been proposed.⁷ The C-Me group of the CH3CHO function is indicated by the respective three proton doublet at 1.60 and 1.65 p.p.m. similar to those of caraparaubine and isoreserpline whose structures (3,4) have recently been reported.⁸ The respective multiplet at 7.30 p.p.m. and 7.28 p.p.m. are assigned to the aromatic protons.

Furthermore, when pteropodine is heated with pyridine, it isomerizes into an equilibrium mixture of pteropodine (20%) and isopteropodine (80%). Similarly, isopteropodine can be isomerized with 10% acetic acid into an equilibrium mixture of pteropodine (70%) and isopteropodine (30%). This characteristic parallel behaviour has also been demonstrated in mitraphylline⁹ and uncarine-A and -B.¹⁰



Since the ultraviolet and infrared spectra are consistent with those of mitraphylline, it is reasonable to suggest that pteropodime and isopteropodime are stereisomeric with mitraphylline and that they have the same structure as (1). Work on the structural degradations of these two alkaloids is in progress.

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